

# CACO-2 Permeability Coefficient Determination in LDTD-MS/MS

P. Tremblay<sup>1</sup>, S. Auger<sup>1</sup>, P. Picard<sup>1</sup>, G. Blachon<sup>1</sup>, B. Julian<sup>2</sup>, L. Laplanche<sup>2</sup>, C. Sarcy<sup>2</sup>, S. Estoul<sup>2</sup>, P. Moliner<sup>2</sup>, O. Fedeli<sup>2</sup> and G. Fabre<sup>2</sup>

<sup>1</sup>Phytronix Technologies, Québec, Canada; <sup>2</sup>Sanofi-Aventis, DSAR, O-C Montpellier, DD, France

Keywords: High-throughput, LDTD, Tandem mass spectrometry, CACO-2 model, Permeability, Hank's buffer

#### Overview

- High-throughput determination of permeability coefficients in early drug discovery development
- CACO-2 / TC-7 model
- Validation against UPLC-MS/MS using 11 commercial compounds

#### Instrumentation

- Phytronix Technologies LDTD ion source (model T-960); Operate with a generic method
- Thermo Fisher Scientific TSQ<sup>®</sup> Vantage<sup>TM</sup> mass spectrometer.

### LDTD ionization process

The LDTD ion source uses an infrared laser diode to desorb sample that have been dried onto a well of a LazWell™ (96-well plate). The desorbed gas phase molecules are carried into a corona discharge region to undergo APCI, and then they are transferred directly into the mass spectrometer for detection.

### **Incubation** (Figure 1)

Apical solution : Hank's balanced Salt Solution (HBSS) pH 7.4 with 0.5 % of Bovine Serum Albumine (BSA) with test compound at 20μM

Basal solution: HBSS pH 7.4 with 5 % of BSA

Incubation performed at 37°C for 2 hours including a pre-incubation of 30 minutes

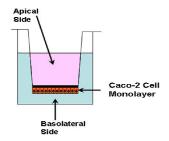


Figure 1 Incubation model

### Sample preparation

Following the incubation period, the Apical solution sample was diluted 1/20 with a HBSS solution at pH 7.4 containing 5 % of BSA. No dilution was performed on the Basal solution sample. The incubation was "stopped" by protein precipitation adding one part of acetonitrile for one part of Basal or Apical solution followed by a centrifugation at 3000 rpm. A calibration curve, ranging from 0.01 to 1  $\mu$ M was performed into HBSS.

The UPLC-MS/MS analysis was performed on the supernatant while a dilution (50/150) with a solution of methanol/water (75/25) containing clomiphene (internal standard) at 50ng/mL was performed for the LDTD-MS/MS analysis.

The permeability coefficients were calculated using the following equation:

$$P_t = C_{Basal} \times V_{Basal} / S \times C_{t=0} \times t$$

Where S = Surface, t the incubation time.

# LDTD-MS/MS analysis

The analysis was performed using a 2  $\mu$ L sample spotted into a 96 LazWell plate. The solvent was evaporated at room temperature.

The carrier gas flow was set at 3 L/min and the laser desorption pattern was the following :

2 seconds at 0 % of Laser power 2 seconds ramping the Laser power up to 45 % 3 seconds plateau at 45 % of Laser power Laser power shut down to 0 % in 0.01 seconds

### Results and Discussion

### LDTD-MS/MS validation with UPLC-MS/MS

11 commercial compounds with a wide range of chemical properties and molecular weights were incubated for permeability study and the permeability coefficient for each compound was obtained following the analysis in UPLC-MS/MS and in LDTD-MS/MS (**Table 1**).

**Table 1** Permeability coefficient (nM/sec) of 11 commercial compounds obtained in LDTD-MS/MS and UPLC-MS/MS

	_				
	•	LDTD		UPLC	
Compound	Formula	Ptot	SD	Ptot	SD
Dextromethorphan	C <sub>18</sub> H <sub>25</sub> NO	172.88	7.97	184.76	0.7
Propranolol	C <sub>16</sub> H <sub>21</sub> NO <sub>2</sub>	216.06	0.00	185.30	30.90
Imipramine	C <sub>19</sub> H <sub>24</sub> N <sub>2</sub>	239.12	16.70	253.30	4.62
Diltiazem	C22H26N2O4S	185.92	6.65	176.23	8.37
Alfusozin	C <sub>19</sub> H <sub>27</sub> N <sub>5</sub> O <sub>4</sub>	1.55	0.50	2.30	0.40
Hydroxyzine	C <sub>21</sub> H <sub>27</sub> ClN <sub>2</sub> O <sub>2</sub>	227.71	21.08	212.71	13.06
Desloratadine	C <sub>19</sub> H <sub>19</sub> ClN <sub>2</sub>	11.79	0.94	14.51	1.02
Atenolol	$C_{14}H_{22}N_2O_3$	3.00	4.20	0.80	0.00
Nadolol	C <sub>17</sub> H <sub>27</sub> NO <sub>4</sub>	0.40	0.00	0.20	0.00
Sulpiride	C <sub>15</sub> H <sub>23</sub> N <sub>3</sub> O <sub>4</sub> S	0.40	0.30	0.40	0.20
Encatropine	$C_{17}H_{25}NO_3$	71.11	3.51	70.60	2.48

The results obtained showed an excellent correlation between both analytical systems with a slope of 1.0225 and r<sup>2</sup> of 0.9862 (**Figure 2**). Moreover, the variability obtained with LDTD-MS/MS is comparable to the one obtained in UPLC-MS/MS.

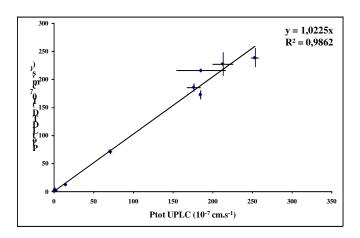


Figure 2 Correlation between LDTD-MS/MS and UPLC-MS/MS for permeability coefficient determination.

# High throughput analysis using LDTD-MS/MS

The LDTD-MS/MS sample-to-sample run time is 25 seconds. This time includes the thermal desorption of the LDTD (10 seconds) and a waiting time coming from the mass spectrometer to be ready for the next sample (15 seconds).

Running the same sample using an UPLC-MS/MS system will takes 4 minutes. Therefore, switching to a LDTD-MS/MS system will increase the sample throughput by a factor of 4.4

#### **Conclusions**

The LDTD-MS/MS allows determining accurate and reproducible permeability coefficient as confirmed by a cross-validation against an UPLC-MS/MS system performed on 11 commercial compounds.

Moreover, the LDTD-MS/MS system allows you to run your samples 4.4 times faster than an UPLC-MS/MS system.

High-throughput analysis with accuracy and precision can be achieved using LDTD as ion source in mass spectrometry.

For more information about your specific application, visit www.phytronix.com

Phytronix Technologies
Parc technologique du Québec métropolitain
4535, boulevard Wilfrid-Hamel, suite 120, Québec (Qc) Canada G1P 2J7
www.phytronix.com